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Bi₂O₃ nano-sheets and nanotubes synthesized by discharges in liquid nitrogen

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Abstract content

Bismuth oxide has a narrower band gap than TiO₂ [1]. It is an excellent photocatalyst for water photolysis and organic pollutants degradation under visible light. As an important type of layer-structured p-type semiconductor, the photocatalytic behaviour of Bi₂O₃ is pretty different from N-type TiO₂. It is also dependent on the allotropic phase as Bi₂O₃ adopts many polymorphic forms, the most common being monoclinic α -phase, tetragonal β -phase, body-centered cubic γ -phase, and face centered cubic δ -phase. In this work, we show how it is possible to synthesize easily Bi₂O₃ nanoparticles by nanosecond pulsed discharges in liquid nitrogen. By creating high-voltage (10 kV) discharges between Bismuth electrodes separated by a distance of 100 μ m typically, Bismuth nanoparticles are produced with three size distributions. Oxidation occurs in a second step, after nitrogen evaporation. Therefore, a part of the nanoparticles exhibit a core-shell structure if oxidation is incomplete. When the applied voltage is lowered to 4-5 kV, ultra-thin nano-sheets and nanotubes also form. The mechanisms responsible for the growth of these new structures are still under investigation but in the case of nanowires, self-catalyzed vapor-liquid-solid route seems highly probable [2, 3]. The microstructure and composition of the nano-objects are characterized by X-ray diffraction, high-resolution TEM (HRTEM) and electron energy loss spectroscopy (EELS). If Bi₂O₃ nanoparticles are amorphous, nano-sheets are cubic, which opens interesting perspectives for their use as active photocatalysts.

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References

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Picture showing the 3 types of Bi₂O₃ nano-objects

